

# 4-*n*-nonylphenol degradation by the genus *Metarhizium* with cytochrome P450 involvement

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## Highlights:

- A novel 4-*n*-NP degradation pathway, distinctive for the genus *Metarhizium* is proposed
- *Metarhizium* species demonstrate differences in the way of 4-*n*-NP derivatives formation
- Cytochrome P450 is involved in the elimination of 4-*n*-NP by *Metarhizium robertsii*

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## Abstract

In this study, the ability of 4-*n*-nonylphenol (4-*n*-NP) elimination by fungal species belonging to the genus *Metarhizium* was investigated. The occurrence of 35 metabolites from 4-*n*-NP degradation was confirmed. For the first time, based on the obtained results, the 4-*n*-NP biodegradation pathway distinctive for the genus *Metarhizium* was proposed. Principal Component Analysis (PCA) indicated that despite the similar elimination pathway in all the examined *Metarhizium* species, there are significant differences in the kinetics of degradation of 4-*n*-NP. Oxidation of the terminal methyl group of the aliphatic chain leading to the formation of carboxylic acids coupled with the removal of terminal carbon is characteristic of *M. robertsii* and *M. guizhouense*, whereas metabolites with a hydroxyl group in the distal part of the nonyl chain distinguish *M. lepidotae* and *M. majus*. Additionally, this study verified the participation of cytochrome P450 in the elimination of the xenobiotic by *Metarhizium* as experimentally proven for *M. robertsii*.

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## 1. Introduction

Endocrine disrupting chemicals (EDCs), chiefly organic compounds, alter the endocrine system of a vertebrate by mimicking natural hormones and by blocking hormone receptors, which causes deleterious effects in humans and in animals even at trace levels (Koumaki et al., 2018; Rajendran et al., 2016; Zheng et al., 2018).

Among all EDCs, 4-nonylphenols (4-NPs) have drawn great attention because of their high estrogenic potential. Exposure of wildlife to EDCs causes multiple harmful effects such as alterations in sexual development, disorders of the homeostasis, delay in growth, and changes in the community structure of the microorganism in the soil (Chang et al., 2007; Janicki et al., 2018; Liu et al., 2017).

The ability of different fungal species to remove NPs has been demonstrated in the literature. To date, biodegradation of NPs by the nonlignolytic fungus, such as *Umbelopsis isabellina* (Janicki et al., 2018), *Metarhizium robertsii* (Różalska et al., 2015b; Szewczyk et al., 2014), *Candida aquatextoris* (Vallini et al., 2001), *Aspergillus* strains (Krupiński et al., 2013; Yang et al., 2018), and other fungal species (Różalska et al., 2010) has been

described. However, to the best of our knowledge, it has not been compared for most species belonging to the same genus. Previously, the ability of the newly isolated *Metarhizium robertsii* IM6519 strain in the degradation of 4-*n*-NP along with the formation of more than 30 derivatives and xenobiotic mineralization was described (Różalska et al., 2015b).

Fungi belonging to the genus *Metarhizium* are unique microorganisms with the ability to parasitize insects. They are abundantly found in the soil and have frequent contact with various anthropogenic pollutants such as NPs. These fungi are often used as biopesticides to control insect populations, and some of the strains are commercially available (Lovett and St. Leger, 2017) and they can also be exploited in the removal of xenobiotics (Szewczyk et al., 2018).

Fungal monooxygenases are involved in various metabolic processes, e.g., in the removal of xenobiotics. They have different structures, have versatile catalytic properties, and have a wide range of substrates (Durairaj et al., 2016). Moreover, monooxygenases engagement in the breakdown of nonylphenol has been also confirmed in the microsomes of

human liver (Deng et al., 2010), Atlantic salmon (Thibaut et al., 2002), as well as in tobacco cells (Berger et al., 2005). Our study has showed that in *Metarhizium* sp. in the presence of the xenobiotic, cytochrome P450 (CYP450) activity increased. Additionally, CYP450 involvement in the elimination of nonylphenol by entomopathogenic fungus was confirmed. Also, for the first time, the universal biodegradation pathway within the genus *Metarhizium* was demonstrated.

## 2. Materials and methods

### 2.1. Strains, cultivation, degradation experiments, and estimation of fungal biomass

The following species of genus *Metarhizium* were cultured: *M. acridum* ARSEF7486, *M. anisopliae* ARSEF7487, *M. brunneum* ARSEF2107, *M. globosum* ARSEF2596, *M. guizhouense* ARSEF6238, *M. lepidiotae* ARSEF7412, *M. majus* ARSEF1914, and *M. robertsii* ARSEF727. These strains were obtained from the Collection of Entomopathogenic Fungal Cultures (ARSEF, USA). The cultures were maintained on ZT agar slants for 14 days and were inoculated in 20 mL Sabouraud medium (Sigma-Aldrich, Germany) in 100 mL Erlenmeyer flasks. After 1 d of incubation, the precultures were transferred to a fresh minimal medium at a ratio of 1:9 (Różalska et al., 2015a). The concentration of 4-*n*-NP in the culture medium was 50 mg L<sup>-1</sup>. Stock solution was prepared in ethanol at a concentration of 20 mg mL<sup>-1</sup>. Abiotic controls (without fungal biomass) and biotic controls (without toxic substrate) were also prepared. The samples were incubated on a rotary shaker (160 rpm) at 28 °C. For the estimation of fungal biomass and for gas chromatography–mass spectrometry (GC–MS) analyses, samples were collected at 6-hour intervals. For biomass estimation the mycelia were filtered through a Whatman filter paper number 1 (Sigma-Aldrich, Germany) and then dried at 105 °C until a constant weight was obtained. Other chemicals and reagents were purchased from Sigma-Aldrich (Germany) unless otherwise stated.

### 2.2. Quantitative analysis of 4-*n*-NP

The fungal cultures were homogenized and extracted with ethyl acetate and methylene chloride (Chempur, Poland) (Różalska et al., 2015b). After evaporation to dryness under reduced pressure, solvent-free residues were dissolved in appropriate amounts of ultra-pure ethyl acetate. The substrate concentration was determined by performing GC–MS (Agilent Technologies, USA, 7890 A series). GC equipped with a mass selective (MS) detector 5975 C and a 30 m HP-5 MS capillary column with an internal diameter of 250 µm and film thickness of 0.25 µm (Agilent Technologies, USA). The injection volume was 2 µL. The inlet was set to a split mode with a split ratio of 10:1 (the split flow was 10 mL min<sup>-1</sup>) and the temperature was maintained at 275 °C. Helium was used as the carrier gas at a constant flow rate of 1.2 mL min<sup>-1</sup>. The column temperature parameters were as follows: 100 °C maintained for 2 min and then increased at a rate of 20 °C min<sup>-1</sup> to 300 °C and maintained

for 8 min. The following temperature program was applied for the mass selective detector: MS source, 250 °C; MS quadrupole, 200 °C; and selective ion monitoring (SIM) mode: target (quantifying) ion at *m/z* 107 and qualifier ion at *m/z* 220. A standard equation was generated for the quantitative analysis in the linear range from 0 to 50 µg mL<sup>-1</sup> 4-*n*-NP (Różalska et al., 2015b).

### 2.3. Qualitative analysis

The metabolites of 4-*n*-NP were identified according to the methods described in previous publications (Krupiński et al., 2014; Różalska et al., 2015b). Briefly, the extracts were dissolved in 1 mL of ethyl acetate and from this, 200 µL sample was transferred into the chromatographic vial. After evaporation to dryness under the N<sub>2</sub> gas stream, 50 µL of *N,O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA) was added and heated to 60 °C for 60 min. Finally, 200 µL of ethyl acetate was added and GC–MS analyses were performed. The injection volume in splitless mode was set to 2 µL. The samples were analyzed with GC–MS at scan mode with the mass range set from 45.0 amu to 550.0 amu. The biodegradation products of 4-*n*-NP were identified as described previously (Różalska et al., 2015b). Moreover, the relative abundances of the metabolites were calculated and the obtained data were subjected to principal component analysis (PCA, MarkerView software version 1.2.1., AB Sciex, USA). Briefly, the orthogonal rotation (varimax) with normalization using total area sums and Pareto scaling on the data set was applied.

### 2.4. Determination of CYP450 activity

CYP450 activity was determined according to the previously published methods with adaptations for human or animal microsomes (Donato et al., 2004; Johnson et al., 2012; Zamaratskaia and Zlabek, 2009). Briefly, microsomes were isolated from the fungal biomass of all the tested *Metarhizium* sp. after 24 h of cultivation in the mineral medium with or without 4-*n*-NP (biotic control). The cultures were filtered through Whatman filter paper number 1 and transferred to the Eppendorf tubes containing glass beads. To each sample, 1 mL of buffer (100 mM potassium phosphate, pH=7.5 containing 1 mM EDTA, 1 mM DTT, 0.5 mM PMSF, 20% glycerol) was added. Disintegration was performed on the FastPrep (MP Biomedicals, USA): speed, 4 m s<sup>-1</sup>; time, 20 s; and cooling on ice, 1 min. After disintegration and centrifugation at 15000 × *g* for 15 min at 4 °C, the supernatant was transferred to an ultra-centrifuge tube and centrifuged again at 105000 × *g* for 1 h 15 min at 4 °C. The microsomal pellet was resuspended in the buffer and the microsomal protein concentrations were assayed with Bradford method using bovine serum albumin as the standard.

CYP450 activities were measured by incubating the prepared microsomes (100 µg of protein) in 150 µL of 100 mM potassium phosphate buffer, pH=7.5, containing the NADPH-regenerating system (500 µM NADPH, 500 µM glucose 6-phosphate and 50

U mL<sup>-1</sup> glucose 6-phosphate dehydrogenase) and 1 mM coumarin as the substrate. After 10 min of incubation, the reaction product (7-hydroxycoumarin) was measured fluorometrically at an excitation wavelength of 355 nm and an emission wavelength of 460 nm in a FLUOstar (BMG Labtech, Germany) microplate reader. In addition, a standard curve using 7-hydroxycoumarin was prepared, and the results of CYP450 activity were expressed as nanomoles of the product per mg of protein.

## 2.5. CYP450 inhibition studies

For CYP450 inhibition studies, the following procedure was adopted. Briefly, 1-aminobenzotriazole (ABT) (1 mM) dissolved in ethanol was added at 0 h and 6 h of incubation to 20 mL of *M. robertsii* ARSEF727 culture supplemented with 4-*n*-NP (50 mg L<sup>-1</sup>) on the mineral medium. After 24 h of incubation, the samples were homogenized and extracted with ethyl acetate and methylene chloride as previously described. In addition, GC–MS qualitative analyses of the metabolites produced were performed after BSFTA derivatization.

## 2.6. Data analysis

Fungal biomass estimation and experiments of CYP450 activity were conducted in six repetitions. Sample variability is given as a standard deviation ( $\pm$  SD). Statistical significance was investigated with the Mann–Whitney *U* test and one-way analysis of variance (ANOVA) (Kruskal–Wallis test) using 4-*n*-NP as the factor on transformed data. Values at *P* < 0.05 were considered as significant. The data were analyzed using Statistica 13.1 (StatSoft, USA). Qualitative data of derivatives were subjected to PCA in orthogonal rotation with normalization using total area sums and Pareto scaling. Marker View software 1.2.1. (AB Sciex, USA) was used to analyze the data.

## 3. Results and discussion

### 3.1. Utilization of 4-*n*-NP by different species of genus *Metarhizium*.

Figure 1 presents the utilization of 4-*n*-NP (50 mg L<sup>-1</sup>) by eight species of the genus *Metarhizium*. All the tested *Metarhizium* sp. showed the ability to remove 4-*n*-NP from the culture medium (see Fig. 1S for additional details). The decrease in the concentration of 4-*n*-NP was observed after 6 h of incubation. Among the tested strains, *M. globosum*, *M. acridum*, *M. guizhouense*, *M. robertsii*, and *M. anisopliae* were the most effective, and after 24 h of incubation, about 15–30% of 4-*n*-NP remained in the culture extracts. At the end of the experiment, less than 5% of 4-*n*-NP was detected. As compared to the abovementioned strains, *M. majus* and *M. lepidiotae* were less efficient in the removal of 4-*n*-NP, in which after 48 h of incubation, 39.3% and 34.5% (respectively) of 4-*n*-NP were detected in the culture medium. Kinetics of degradation of 4-*n*-NP for *M. brunneum* was found to be completely different from that of the abovementioned strains. In this case, after 24 h of incubation, 55% of 4-*n*-NP was detected, but after 48 h of incubation, it was below 10%. The amount of 4-*n*-NP in abiotic controls remained constant during this experiment.

In our previous study, we demonstrated that *M. robertsii* IM6519 can degrade and mineralize 4-*n*-NP (Różalska et al., 2015b). The strain removed 85% of the 4-*n*-NP after 24 h of incubation (initial concentration of 40 mg L<sup>-1</sup>) (Różalska et al., 2015b). In this study, *M. robertsii* ARSEF727 also removed 85% of 4-*n*-NP (initial concentration of 50 mg L<sup>-1</sup>) (Fig. 1). This shows the similarity in both the strains of *M. robertsii*. However in this study we further show that biodegradation of this xenobiotic was found to be characteristic of the genus *Metarhizium*.

In the literature, there are studies describing the ability of filamentous fungi, ligninolytic and nonligninolytic, to degrade 4-

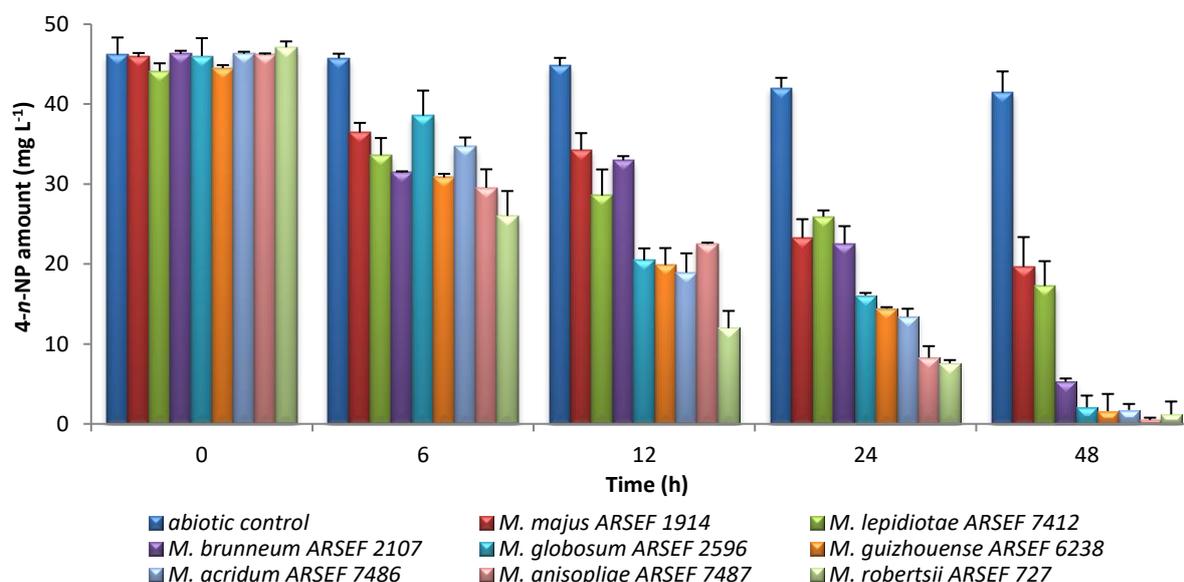
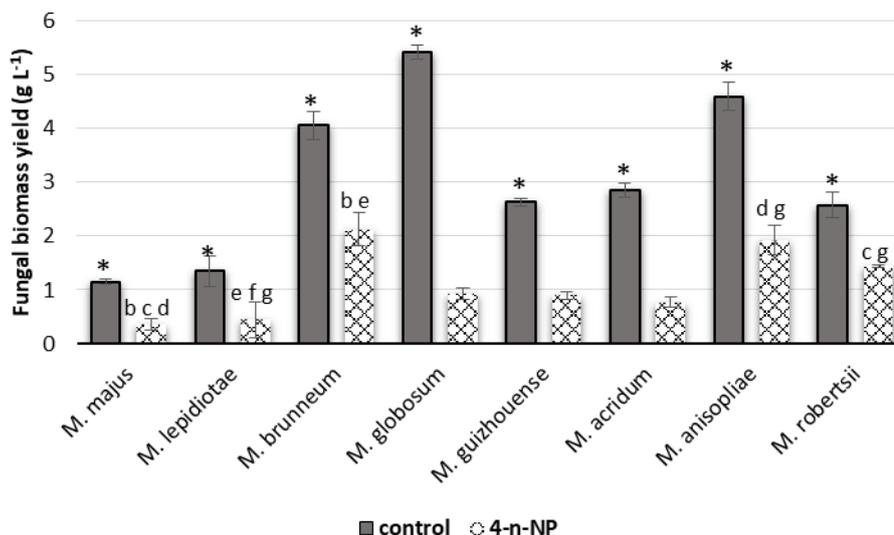


Fig. 1. 4-*n*-nonylphenol (4-*n*-NP) utilization by species belonging to the genus *Metarhizium*.



**Fig. 2.** 4-*n*-nonylphenol (4-*n*-NP) influence on biomass yield of *Metarhizium* sp. after 1 d of incubation. (\* $P < 0.05$  – statistically significant decrease in biomass in the sample with 4-*n*-NP relative to the control). Results were tested by one-way ANOVA. Statistical significance was also noticed between samples with 4-*n*-NP: *Metarhizium majus* and *Metarhizium brunneum* (b,  $P = 0.000552$ ), *M. majus* and *Metarhizium robertsii* (c,  $P = 0.000760$ ), *M. majus* and *Metarhizium anisopliae* (d,  $P = 0.003368$ ), *Metarhizium lepidiotae* and *M. brunneum* (e,  $P = 0.001554$ ), *M. lepidiotae* and *M. robertsii* (f,  $P = 0.002108$ ), and *M. lepidiotae* and *M. anisopliae* (g,  $P = 0.008625$ ).

*n*-NP, but to the best of our knowledge, there are no such papers describing the degradation potential on the level of one genus. Therefore, it is very difficult to draw coherent conclusions regarding the degradation capacity within one genus because of different incubation times and various concentrations of the xenobiotics tested (Cabana et al., 2007; Cajthaml, 2014; Corvini et al., 2006).

### 3.2. Biomass content of *Metarhizium* sp. in the presence of 4-*n*-NP

Figure 2 shows that in the case of all the tested *Metarhizium* sp., the biomass content was lower in the presence of 4-*n*-NP than that of control samples ( $P < 0.05$ ). The highest growth inhibition in the presence of 4-*n*-NP was demonstrated for *M. globosum* and then for *M. acridum* (83% and 73%, respectively, with reference to the control) ( $P < 0.05$ ). In *M. majus*, *M. lepidiotae*, and *M. guizhouense*, the biomass obtained from the samples with 4-*n*-NP was about 70% lower than that of the control ( $P < 0.05$ ). It is noteworthy that for *M. majus* and *M. lepidiotae*, the amount of biomass in biotic controls was the lowest among the tested species. For the other strains, the difference in the amount of biomass between biotic and tested samples was less visible, although statistically significant.

### 3.3. *Metarhizium* sp. degradation pathway

Qualitative analyses of the metabolites formed during the degradation of 4-*n*-NP by the eight tested species of the genus *Metarhizium* were based on the samples collected at five time points—at 0 h and after 6, 12, 24, and 48 h of incubation. The degradation products of 4-*n*-NP were identified based on the characteristic fragmentation patterns described in our previous study (Różalska et al., 2015b). The data were confirmed with

the NIST08 Mass Spectral Database with a probability range of 90–99% (Tab. 1).

Figure 3 shows the proposed pathway of 4-*n*-NP degradation by tested *Metarhizium* species, which is very complex. All xenobiotic derivatives as listed in Table 1 were detected in the culture extracts of these strains. PCA was used to evaluate the complex results. Figure 4 shows PC1/PC2 scores and loading plots acquired by performing orthogonal rotation PCA, with normalization using total area sums and Pareto scaling.

The arrangement of the tested species on the PCA chart (Fig. 4) corresponds, to a large extent, with the data on the elimination of 4-*n*-NP by *Metarhizium* sp. (Fig. 1). All the derivatives were detected in the extracts of the eight tested species suggesting that the degradation pathway of 4-*n*-NP is a universal type for *Metarhizium* sp. However, we observed significant differences in the kinetics of formation of these derivatives among the tested strains. It was found that at particular stages of xenobiotic degradation, the tested species showed the presence of derivatives of various parts of the proposed pathway of 4-*n*-NP degradation. Therefore, the degradation pathway was divided into four groups (groups I–IV) of derivatives, which allows for a better understanding of the individual stages of 4-*n*-NP degradation among the tested species. The derivatives assigned to group I possess a hydroxyl group next to the aromatic ring in the nonyl-moiety and some of them also have the second (versatile) hydroxyl group in the nonyl chain. Group II derivatives include monohydroxylated metabolites with a hydroxyl group in the distal part of the nonyl chain. The tested *Metarhizium* species were capable of consecutively oxidizing the terminal methyl group of the aliphatic chain, leading to the formation of carboxylic acids coupled with terminal removal of carbon. This is the most common mechanism of 4-*n*-NP biodegradation by microorga-

**Tab. 1.** Gas chromatography–mass spectroscopy results of qualitative analysis of 4-*n*-nonylphenol (4-*n*-NP) degradation by *Metarhizium* sp.

ID	Retention time (min)	Compound name (TMS-derivatives)	Chemical formula	Molecular mass (Da)	Mass spectrum <i>m/z</i> (10 largest ions relative intensity)
1	7.520	Phenylacetic acid, TMS	C <sub>11</sub> H <sub>16</sub> O <sub>2</sub> Si	208.32	73 (99.9), 75 (34), 91 (19.7), 164 (19.7), 193 (12.5), 45 (9.9), 65 (9.9), 74 (9.7), 89 (5.8), 90 (5.7)
2	8.713	1-(4-hydroxyphenyl)ethanone, TMS	C <sub>11</sub> H <sub>16</sub> O <sub>2</sub> Si	208.33	193 (99.9), 208 (25.6), 194 (16.3), 73 (14.6), 151 (12.9), 45 (10.2), 75 (5), 89 (7.4), 91 (5.5), 195 (4.8)
3	9.652	4-Hydroxybenzoic acid, di-TMS	C <sub>13</sub> H <sub>22</sub> O <sub>3</sub> Si <sub>2</sub>	282.49	267 (99.9), 223 (73.6), 193 (58), 73 (50.9), 282 (24.4), 268 (23.5), 224 (16.4), 126 (11.8), 194 (10.3), 269 (9.8)
4	9.716	2-(4-Hydroxyphenyl)acetic acid, di-TMS	C <sub>14</sub> H <sub>24</sub> O <sub>3</sub> Si <sub>2</sub>	296.52	73 (99.9), 179 (30.1), 75 (16.1), 164 (15.8), 252 (15.7), 296 (915.3), 281 (14.7), 74 (9), 45 (8.5), 163 (6.6)
5	10.411	3-(4-Hydroxyphenyl)propanoic acid, di-TMS	C <sub>15</sub> H <sub>26</sub> O <sub>3</sub> Si <sub>2</sub>	310.54	179 (99.9), 192 (63.3), 73 (29.3), 235 (24.6), 310 (22.4), 177 (17.4), 193 (16.9), 180 (15.6), 75 (13.1), 295 (7.2)
6	10.74	3,4-Dihydroxybenzoic acid, tri-TMS	C <sub>16</sub> H <sub>30</sub> O <sub>4</sub> Si <sub>3</sub>	370.67	193 (99.9), 370 (55.7), 73 (48.9), 355 (29.3), 311 (20.1), 371 (18.5), 194 (15.2), 281 (10.9), 223 (9.7), 356 (9.4)
7	10.798	(3,4-Dihydroxyphenyl)acetic acid, tri-TMS	C <sub>17</sub> H <sub>32</sub> O <sub>4</sub> Si <sub>3</sub>	384.69	73 (99.9), 179 (60.9), 384 (55.5), 267 (49.7), 385 (19.3), 237 (15.8), 75 (13.9), 268 (13.7), 45 (9.1), 369 (9.1)
8	10.962	4-(4-Hydroxyphenyl)butanoic acid, di-TMS	C <sub>16</sub> H <sub>28</sub> O <sub>3</sub> Si <sub>2</sub>	324.57	192 (99.9), 73 (24.1), 193 (18.6), 179 (17.9), 309 (16.2), 324 (15.7), 177 (15.4), 75 (14.7), 147 (7.6), 194 (5.1)
9	11.249	4- <i>n</i> -Nonylphenol, TMS	C <sub>18</sub> H <sub>32</sub> O <sub>2</sub> Si	292.54	179 (99.9) 292 (39) 180 (21.8) 73 (16.6), 293 (10.6) 181 (6.6) 277 (5.9) 163 (4.0), 165 (3.7) 149 (3.3)
10	11.346	(2 <i>E</i> )-3-(4-Hydroxyphenyl)prop-2-enoic acid, di-TMS	C <sub>15</sub> H <sub>24</sub> O <sub>3</sub> Si <sub>2</sub>	308.52	293 (99.9), 219 (85.2), 308 (80), 73 (55.2), 249 (48.8), 294 (25.8), 309 (21.4), 220 (18.9), 179 (14.8), 250 (10.6)
11	11.395	3-(3,4-Dihydroxyphenyl)propanoic acid, tri-TMS	C <sub>18</sub> H <sub>34</sub> O <sub>4</sub> Si <sub>3</sub>	398.71	179 (99.9), 398 (91.5), 267 (55.2), 73 (51.9), 399 (31.3), 280 (17.5), 180 (14.6) 400 (14.2), 268 (13.6), 383 (12.1)
12	11.464	5-(4-Hydroxyphenyl)pentanoic acid, di-TMS	C <sub>17</sub> H <sub>30</sub> O <sub>3</sub> Si <sub>2</sub>	338.60	179 (99.9), 73 (43.4), 75 (24.1), 338 (23.8), 192 (23.2), 323 (19.3), 180 (16.3), 149 (15.9), 205 (13.4), 206 (7.9)
13	11.858	4-(3,4-Dihydroxyphenyl)butanoic acid, tri-TMS	C <sub>19</sub> H <sub>36</sub> O <sub>4</sub> Si <sub>3</sub>	412.74	412 (99.9), 280 (99.5), 73 (78.1), 267 (37.7), 413 (35), 235 (29.7), 281 (26.8), 179 (25.7), 397 (18.3), 75 (18)
14	11.925	4-Hydroxy-4-(4-hydroxyphenyl)butanoic acid, tri-TMS	C <sub>19</sub> H <sub>36</sub> O <sub>4</sub> Si <sub>3</sub>	412.74	412 (99.9), 73 (81.2), 267 (60.4), 179 (47.9), 413 (37.2), 268 (16.3), 414 (16.1), 233 (15.5), 75 (10.3), 74 (7.2)
15	11.965	6-(4-Hydroxyphenyl)hexanoic acid, di-TMS	C <sub>18</sub> H <sub>32</sub> O <sub>3</sub> Si <sub>2</sub>	352.63	179 (99.9) 352 (31.7) 73 (31) 180 (16.1) 337 (14.9) 75 (14.7), 353 (9.4) 253 (6.9) 181 (4.6), 177 (4.5)
16	12.127	4-(1-Hydroxynonyl)phenol, di-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>2</sub> Si <sub>2</sub>	380.71	380 (99.9), 267 (82), 73 (76.7), 179 (37.5), 381 (34.8), 268 (23.3), 382 (12.8), 269 (8.9), 365 (7.9), 180 (6.9)
17	12.266	8-(4-Hydroxyphenyl)nonanal, TMS	C <sub>18</sub> H <sub>30</sub> O <sub>2</sub> Si	306.51	179 (99.9), 306 (26.2), 73 (22), 180 (16.4), 43 (8.8), 181 (7.5), 307 (6.6), 163 (3.8), 75 (3.7), 149 (3.5)
18	12.237	7-(4-Hydroxyphenyl)nonanal, TMS	C <sub>18</sub> H <sub>30</sub> O <sub>2</sub> Si	306.51	179 (99.9), 306 (26.4), 73 (23), 180 (16), 181 (7.5), 57 (7.3), 307 (6.5), 192 (5.7), 75 (4.1), 177 (4)
19	12.285	(2 <i>E</i> )-3-(3,4-Dihydroxyphenyl)prop-2-enoic acid, tri-TMS	C <sub>18</sub> H <sub>32</sub> O <sub>4</sub> Si <sub>3</sub>	396.70	219 (99.9), 73 (83.7), 396 (75), 397 (27.1), 381 (19), 191 (17.3), 220 (17.2), 218 (15.2), 179 (14.9), 398 (12.3)
20	12.316	5-(3,4-Dihydroxyphenyl)pentanoic acid, tri-TMS	C <sub>20</sub> H <sub>38</sub> O <sub>4</sub> Si <sub>3</sub>	426.76	426 (99.9), 73 (94.2), 267 (77.7), 427 (37), 205 (31.7), 75 (23), 268 (21), 411 (20.9), 179 (20.3), 428 (16.9)
21	12.373	5-Hydroxy-5-(4-hydroxyphenyl)pentanoic acid, tri-TMS	C <sub>20</sub> H <sub>38</sub> O <sub>4</sub> Si <sub>3</sub>	426.76	73 (99.9), 426 (80.6), 267 (51.4), 179 (47.9), 427 (31.7), 75 (19.9), 268 (15), 253 (14.9), 428 (14), 147 (6.1)
22	12.373	4-(6-Hydroxynonyl)phenol, di-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>2</sub> Si <sub>2</sub>	380.71	73 (99.9), 179 (90.2), 290 (67.8), 145 (58.9), 205 (57.1), 75 (31), 291 (17.2), 192 (15.5), 180 (15.1), 206 (11.5)
23	12.416	7-(4-Hydroxyphenyl)heptanoic acid, di-TMS	C <sub>19</sub> H <sub>34</sub> O <sub>3</sub> Si <sub>2</sub>	366.65	179 (99.9), 73 (41.9), 366 (28), 75 (17.6), 180 (16.2), 351 (10.4), 367 (8.8), 253 (6.6), 181 (5.1), 177 (5)
24	12.523	4-(7-Hydroxynonyl)phenol, di-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>2</sub> Si <sub>2</sub>	380.71	179 (99.9), 73 (85.7), 131 (84.2), 290 (51.8), 205 (34.4), 75 (24.7), 351 (20.5), 380 (19.6), 180 (16.5), 291 (13)
25	12.588	4-(8-Hydroxynonyl)phenol, di-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>2</sub> Si <sub>2</sub>	380.71	179 (99.9), 117 (97.3), 73 (76.1), 290 (30), 75 (27.8), 380 (26.2), 205 (19.2), 180 (16.8), 118 (10.7), 365 (9.2)
26	12.753	6-Hydroxy-6-(4-hydroxyphenyl)hexanoic acid, tri-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>4</sub> Si <sub>3</sub>	440.79	73 (99.9), 440 (59.3), 179 (49.8), 267 (45.4), 75 (29.4), 441 (22.9), 197 (14.9), 268 (12.6), 442 (10.1), 180 (7.9)
27	12.796	6-(3,4-Dihydroxyphenyl)hexanoic acid, tri-TMS	C <sub>21</sub> H <sub>40</sub> O <sub>4</sub> Si <sub>3</sub>	440.79	73 (99.9), 440 (99.7), 267 (80), 179 (46.6) 441 (40), 268 (21.6), 75 (20.1), 253 (18.8), 442 (18), 87 (14.4) 269 (8.3)
28	12.903	(4-Hydroxyphenyl)octanoic acid, di-TMS	C <sub>20</sub> H <sub>36</sub> O <sub>3</sub> Si <sub>2</sub>	380.66	179 (99.9), 73 (59.2), 380 (28.7), 180 (16.7), 75 (12.4), 381 (9.9), 253 (8.2), 103 (6.8), 181 (6.5), 147 (6.1)
29	13.104	4-(1,6-Dihydroxynonyl)phenol, tri-TMS	C <sub>24</sub> H <sub>48</sub> O <sub>3</sub> Si <sub>3</sub>	468.89	73 (99.9), 468 (32.5), 179 (29.1), 267 (28.3), 145 (27.2), 75 (20.7), 469 (13.7), 378 (12), 293 (10.3), 268 (8.3)
30	13.169	7-Hydroxy-7-(4-hydroxyphenyl)heptanoic acid, tri-TMS	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub> Si <sub>3</sub>	454.82	454 (99.9), 73 (81.5), 267 (73.6), 179 (57.6), 455 (39.4), 268 (20.4), 75 (19.5), 439 (19.5), 456 (17.9), 180 (9.3)
31	13.247	4-(1,7-Dihydroxynonyl)phenol, tri-TMS	C <sub>24</sub> H <sub>48</sub> O <sub>3</sub> Si <sub>3</sub>	468.89	73 (99.9), 468 (57.7), 267 (41), 131 (40.7), 179 (28.7), 469 (23.7), 75 (19), 268 (10.9), 253 (10.4), 439 (10.3)

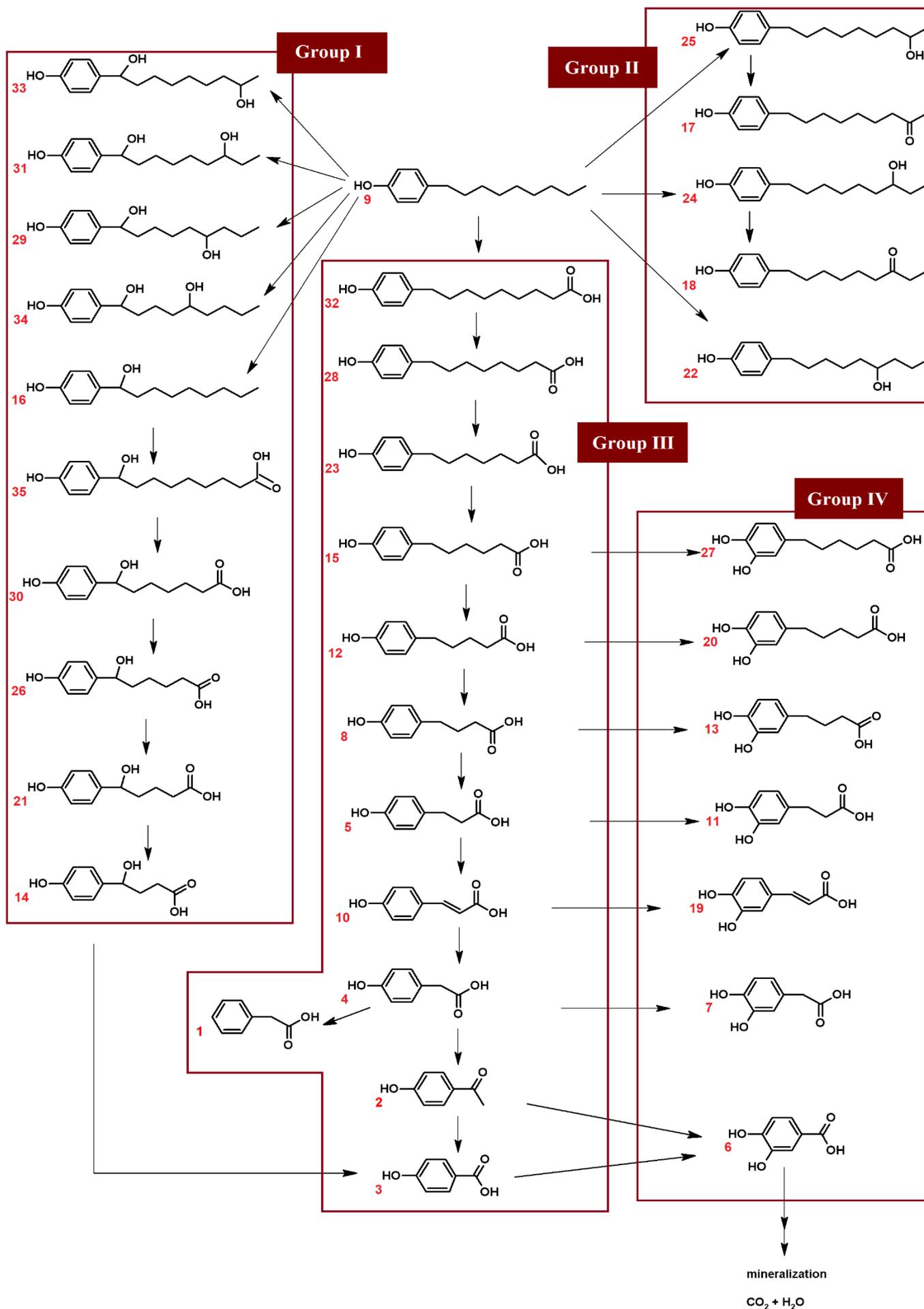
32	13.283	9-(4-Hydroxyphenyl)nonanoic acid, di-TMS	C <sub>21</sub> H <sub>38</sub> O <sub>3</sub> Si <sub>2</sub>	394.69	179 (99.9), 73 (41.9), 394 (28), 180 (16.5), 75 (16), 379 (11.5), 395 (9.2), 181 (5.2), 177 (4.6), 253 (4.6)
33	13.311	4-(1,8-Dihydroxynonyl)phenol, tri-TMS	C <sub>24</sub> H <sub>48</sub> O <sub>3</sub> Si <sub>3</sub>	468.89	73 (99.9), 468 (75.3), 117 (59), 267 (48.6), 179 (32), 469 (31.9), 75 (22.7), 268 (21.8), 211 (18), 253 (15.1)
34	13.605	4-(1,5-Dihydroxynonyl)phenol, tri-TMS	C <sub>24</sub> H <sub>48</sub> O <sub>3</sub> Si <sub>3</sub>	468.89	73 (99.9), 468 (76.7), 267 (41.8), 179 (35.9), 469 (31), 75 (25.3), 470 (14.5), 411 (12.3), 268 (11.6), 221 (10.6)
35	13.97	9-Hydroxy-9-(4-hydroxyphenyl)nonanoic acid, tri-TMS	C <sub>24</sub> H <sub>46</sub> O <sub>4</sub> Si <sub>3</sub>	482.27	73 (99.9), 482 (88.9), 267 (54), 179 (49.4), 483 (36.9), 75 (22.8), 484 (18.1), 467 (18), 268 (14.9), 74 (9.2)

nisms (group III derivatives) (Krupiński et al., 2013; Rajendran et al., 2016; Vallini et al., 2001). Group (IV) derivatives are formed by the hydroxylation of the aromatic ring, and they are probably formed from the derivatives belonging to group III. It was observed that each of the tested species showed a preference for the degradation of 4-*n*-NP to a particular group of derivatives.

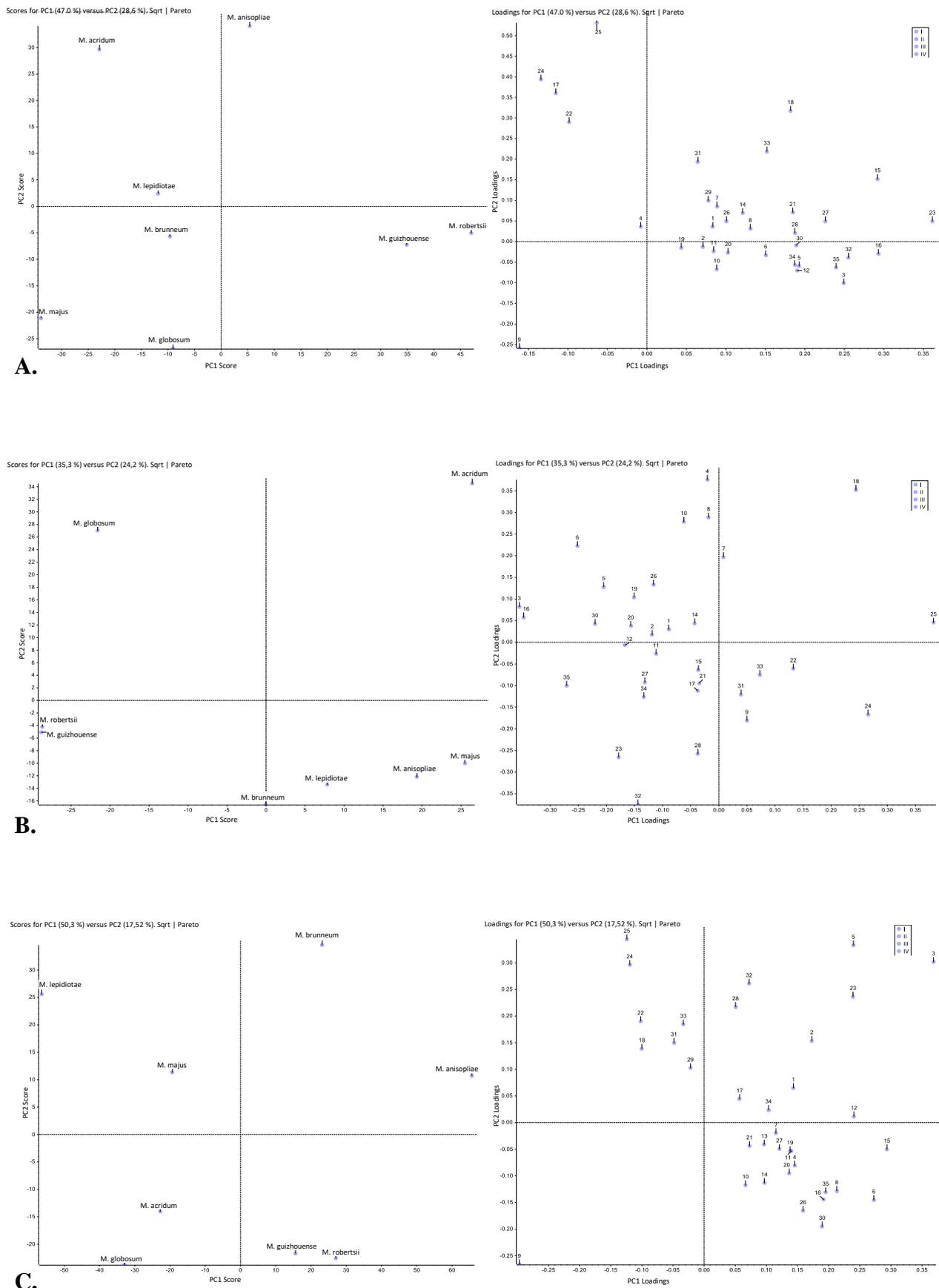
Based on the results from the PCA analysis, at 6, 12, and 24 h of incubation, a pronounced similarity between *M. guizhouense* and *M. robertsii* strains was observed in the formation of derivatives of 4-*n*-NP degradation (Fig. 4). The trend analysis of product formation revealed that derivatives belonging to groups III and IV were definitely predominant in *M. guizhouense* and *M. robertsii* culture extracts. In fact, all the derivatives belonging to group III were detected after 6 h of incubation. The derivatives 9-(4-Hydroxyphenyl)nonanoic acid, di-trimethylsilyl (TMS) (**32**) and (4-Hydroxyphenyl)octanoic acid, di-TMS (**28**) from the initial part of the pathway were detected at high levels. Derivatives of the end of the route for group III such as 1-(4-Hydroxyphenyl)ethanone, TMS (**2**); 4-Hydroxybenzoic acid, di-TMS (**3**); 2-(4-Hydroxyphenyl)acetic acid, di-TMS (**4**); and (2E)-3-(4-Hydroxyphenyl)prop-2-enoic acid, di-TMS (**10**) were detected as soon as after 6 h incubation and consistently appeared at high levels at subsequent incubation times. Their content was the lowest at 24 h of incubation. This might be due to the fact that after 24 h of incubation, only 5% of the substrate remained in the culture medium.

In *M. guizhouense* cultures, all derivatives belonging to group IV (with hydroxylation in the aromatic ring) were detected at very high levels as soon as after 6 h of incubation. This indicates fast kinetics of degradation by this strain. In *M. robertsii*, these derivatives were also detected but at lower levels (Fig. 4A). 6-(3,4-Dihydroxyphenyl)hexanoic acid, tri-TMS (**27**), derived from 6-(4-Hydroxyphenyl)hexanoic acid, di-TMS (**15**) (located in the middle of the route of group III) was also detected at a very high level. In *M. robertsii*, at 6 h of incubation, the decomposition of metabolites belonging to group III was found to be initiated, which suggested its lower kinetics of 4-*n*-NP degradation than that observed for *M. guizhouense*. At 12 and 24 h of incubation, the percentage of derivatives belonging to group IV remained at a similar high levels in both *M. guizhouense* and *M. robertsii*. These species also exhibited the degradation of 4-*n*-NP to group I metabolites (a hydroxyl group next to the aromatic ring in the nonyl moiety or/and the versatile second hydroxylation in the nonyl chain). Derivatives such as 4-(1,5-Dihydroxynonyl)phenol, tri-TMS (**34**) or 4-(1,8-Dihydroxynonyl)phenol, tri-TMS (**33**) and also 4-(1-

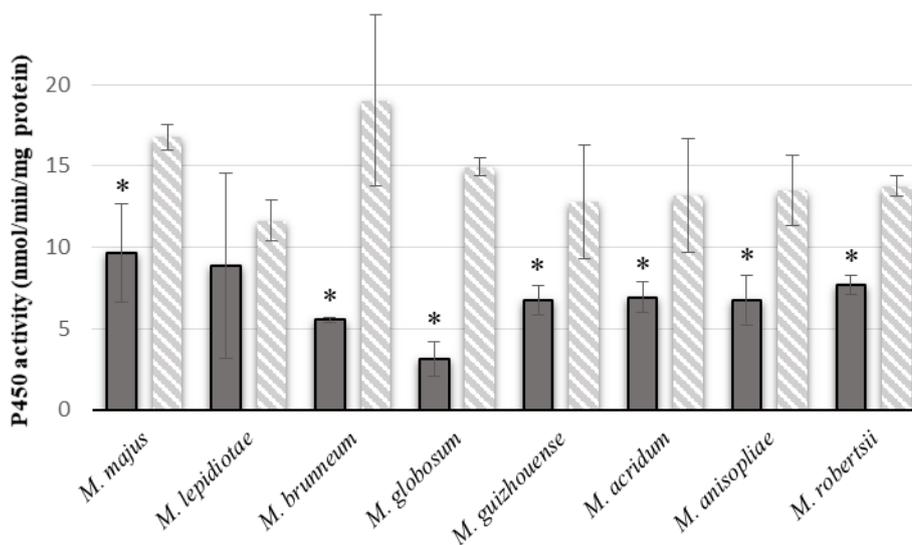
Hydroxynonyl)phenol, di-TMS (**16**) (with a hydroxyl group alone next to the aromatic ring), and their degradation metabolites, e.g., 5-Hydroxy-5-(4-hydroxyphenyl)pentanoic acid, tri-TMS (**21**) and 4-Hydroxy-4-(4-hydroxyphenyl)butanoic acid, tri-TMS (**14**) were detected as soon as after 6 h of incubation. In the following periods of incubation, these derivatives were detected as well, but differences were observed in their mutual ratio. In the case of *M. guizhouense* and *M. robertsii*, the following metabolites belonging to group II were detected at very low levels at 6, 12, and 24 h of incubation: 7-(4-Hydroxyphenyl)nonanal, TMS (**18**) formed from 4-(7-Hydroxynonyl)phenol, di-TMS (**24**) at 6 h of incubation and 8-(4-Hydroxyphenyl)nonanal, TMS (**17**) formed from 4-(8-Hydroxynonyl)phenol, di-TMS (**25**) at 12 h of incubation. There is a possibility that these strains degraded 4-*n*-NP into derivatives belonging to group II already at the beginning of the incubation; therefore, at later periods of incubation, there was a low level of derivatives belonging this group. This part of the pathway was not dominant for these two strains. At 24 h of incubation, the derivatives belonging to group II were not detected for *M. guizhouense* and *M. robertsii*. *M. lepidiotae* and *M. majus* exhibited large similarities in the degradation of 4-*n*-NP (Fig. 4). The trend analyses of the nascent compounds revealed that group II of the pathway was definitely the main degradation route for these strains, however, *M. lepidiotae* showed the presence of more derivatives than *M. majus* in their culture extracts. At 6 h of incubation, the efficiency of 4-*n*-NP degradation in these two strains was at the lowest level among all tested species (Fig. 1). However, in *M. lepidiotae*, the derivatives with hydroxylation at the C2–C4 alkyl side-chain of 4-*n*-NP (group II), i.e., (8-Hydroxynonyl)phenol, di-TMS (**25**); 4-(7-Hydroxynonyl)phenol, di-TMS (**24**); and 4-(6-Hydroxynonyl)phenol, di-TMS (**22**) were detected at levels of 40–60%, whereas in *M. majus*, they were detected at levels of 15–25%. The percentage of these derivatives increased up to 60–100% at 12 h of incubation. In *M. lepidiotae*, at 6 and 12 h of incubation, the derivatives with alkyl-chain hydroxylation next to the aromatic ring (group I) were detected, such as 4-(1,8-Dihydroxynonyl)phenol, tri-TMS (**33**); 4-(1,7-Dihydroxynonyl)phenol, tri-TMS (**31**); and 4-(1-Hydroxy-nonyl)phenol, di-TMS (**16**) as well as derivatives of the beginning of the path for group III, such as 9-(4-Hydroxyphenyl)nonanoic acid, di-TMS (**32**) and (4-Hydroxyphenyl)octanoic acid, di-TMS (**28**). However, in *M. majus*, the formation of these metabolites began as late as after 24 h of incubation. Among *M. majus* and *M. lepidiotae*, *M.*



**Fig. 3.** Proposed 4-n-nonylphenol (4-n-NP) degradation pathway by tested species of the genus *Metarhizium* with division into four groups (I-IV) of derivatives. The red numbers refer to the metabolites shown in Table 1 (see text for details). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)



**Fig. 4.** Results of principal component analysis (PCA) on the 4-*n*-nonylphenol (4-*n*-NP) metabolism by the eight species from genus *Metarhizium*. On the right—PC1 against PC2 loadings chart; on the left—PC1 against PC2 scores chart. **A.** PCA chart at 6 h of incubation. **B.** PCA chart at 12 h of incubation. **C.** PCA chart at 24 h of incubation (see Supplementary material for additional details—Fig. 3S–5S).



**Fig. 5.** Cytochrome P450 activity in microsomes of *Metarhizium* sp. in the absence (control) and in the presence of 4-*n*-nonylphenol (4-*n*-NP) (\**P* < 0.05—a statistically significant increase in cytochrome P450 activity in the sample with 4-*n*-NP relative to the control).

*lepidiotae* degraded 4-*n*-NP with higher efficiency than that of *M. majus* at the end of incubation (Fig. 1), although the way of derivatives formation in both strains was similar. It is noteworthy that *M. majus* and *M. lepidiotae* were the slowest growing species within the genus *Metarhizium* (Fig. 2).

*M. globosum* is presented as a separate issue, due to the lack of significant similarity with the other *Metarhizium* species. In this strain, derivatives belonging to group III dominated throughout the incubation period. At 6 h of incubation, 4-*n*-NP degradation was less efficient compared to the other strains (Fig. 1). The trend analysis showed that the breakthrough of the degradation process of 4-*n*-NP for *M. globosum* occurred at 12 h of incubation. The amount of derivatives belonging to group III, in particular, to those from the end of the path, was at the highest level. In addition, all derivatives belonging to group IV (**6**, **7**, **11**, **19**, **20**, **27**) with aromatic ring hydroxylation were detected. At 12 h of incubation, a high percentage of derivatives with alkyl-chain hydroxylation next to the aromatic ring (group I), in particular 4-(1-Hydroxynonyl)phenol, di-TMS (**16**); 6-Hydroxy-6-(4-hydroxyphenyl)hexanoic acid, tri-TMS (**26**); and 4-Hydroxy-4-(4-hydroxyphenyl)butanoic acid, tri-TMS (**14**) was detected. After 24 h of incubation, the kinetics of 4-*n*-NP degradation decreased and there was no distinctive derivative detected. The very low content of metabolites from group II throughout the incubation period was found to be distinctive for this strain. *M. brunneum* showed some interesting results. The dominant products of 4-*n*-NP degradation for this strain were metabolites belonging to group III, similar to *M. globosum*. However, *M. brunneum* definitely differed in the kinetics of 4-*n*-NP degradation. After 6 and 12 h of incubation, the derivatives of the initial part of the path of group III were detected, such as 9-(4-Hydroxyphenyl)nonanoic acid, di-TMS (**32**) and (4-Hydroxyphenyl)octanoic acid, di-TMS (**28**). In contrast to *M. globosum*, in *M. brunneum*, the degradation of 4-*n*-NP to derivatives of the end of the path for group III began as late as after 24 h of incubation and metabolites such as Phenylacetic acid, TMS (**1**); 4-Hydroxybenzoic acid, di-TMS (**3**);

and 3-(4-Hydroxyphenyl)propanoic acid, di-TMS (**5**) were detected at very high levels. Derivatives belonging to groups I, II, and IV were also detected but at very low levels.

*M. acridum* and *M. anisopliae* were found to be similar in the kinetics of formation of 4-*n*-NP derivatives at 6 h of incubation. The trend analyses revealed that for both these strains, hydroxylation in the C2–C4 alkyl side-chain of 4-*n*-NP (group II) predominated. In *M. anisopliae*, high levels of 4-(8-Hydroxynonyl)phenol, di-TMS (**25**); 4-(7-Hydroxynonyl)phenol, di-TMS (**24**); and 4-(6-Hydroxynonyl)phenol, di-TMS (**22**) and low levels of 8-(4-Hydroxyphenyl)nonanal, TMS (**17**) and 7-(4-Hydroxyphenyl)nonanal, TMS (**18**) were detected. *M. acridum* showed contrasting results. Most likely, in *M. acridum*, the metabolites: 4-(8-Hydroxynonyl)phenol, di-TMS (**25**) to 8-(4-Hydroxyphenyl)nonanal, TMS (**17**) and 4-(7-Hydroxynonyl)phenol, di-TMS (**24**) to 7-(4-Hydroxyphenyl)nonanal, TMS (**18**) were detected earlier than in *M. anisopliae*. Derivatives belonging to group III were also detected, but differentiation between *M. acridum* and *M. anisopliae* occurred with reference to derivatives with a hydroxyl group next to the aromatic ring in the nonyl-moiety or/and the versatile second hydroxylation in the nonyl chain (I group). In *M. acridum*, derivatives, such as 4-(1,8-Dihydroxynonyl)phenol, tri-TMS (**33**) and 4-(1,7-Dihydroxynonyl)phenol, tri-TMS (**31**) were detected at very low levels (below 20%). However, in *M. anisopliae*, these derivatives including the others, e.g., 4-(1-Hydroxynonyl)phenol, di-TMS (**16**) and 5-Hydroxy-5-(4-hydroxyphenyl)pentanoic acid, tri-TMS (**21**) were dominant (55–100%). A significant difference between these strains was observed at 12 h of incubation (Fig. 4B). Only metabolites belonging to group II were found to be similar. In *M. anisopliae*, the degradation of 4-*n*-NP occurred mainly in the part of the pathway with hydroxylation next to the aromatic ring in the nonyl-moiety or/and the versatile second hydroxylation in nonyl chain (I group), whereas in *M. acridum*, the degradation was observed mainly in the part of the pathway with carboxylic

**Tab. 2.** Gas chromatography–mass spectroscopy results of qualitative analysis of 4-*n*-nonylphenol (4-*n*-NP) degradation by *Metarhizium* sp.

Group of derivatives	ID	Compound name (TMS-derivative)	Samples without addition of ABT	Samples with addition of ABT at 0h of incubation	Samples with addition of ABT after 6h of incubation
I	16	4-(1-hydroxynonyl)phenol, di-TMS	✓	✗	✓
	26	6-hydroxy-6-(4-hydroxyphenyl) hexanoic acid, tri-TMS	✓	✗	✓
	30	7-hydroxy-7-(4-hydroxyphenyl) heptanoic acid, tri-TMS	✓	✗	✓
	35	9-Hydroxy-9-(4-hydroxyphenyl) nonanoic acid, tri-TMS	✓	✗	✓
III	1	Phenylacetic acid, TMS	✓	✓	✓
	2	1-(4-hydroxyphenyl)ethanone, TMS	✓	✗	✓
	3	4-Hydroxybenzoic acid, di-TMS	✓	✓	✓
	4	2-(4-Hydroxyphenyl)acetic acid, di-TMS	✓	✓	✓
	5	3-(4-Hydroxyphenyl)propanoic acid, di-TMS	✓	✗	✓
	8	4-(4-hydroxyphenyl)butanoic acid, di-TMS	✓	✗	✓
	10	(2E)-3-(4-hydroxyphenyl)prop-2-enoic acid, di-TMS	✓	✗	✓
	12	5-(4-hydroxyphenyl)pentanoic acid, di-TMS	✓	✗	✓
	15	6-(4-hydroxyphenyl)hexanoic acid, di-TMS	✓	✗	✓
	32	9-(4-hydroxyphenyl)nonanoic acid, di-TMS	✓	✗	✓
IV	6	3,4-Dihydroxybenzoic acid, tri-TMS	✓	✗	✓
	20	5-(3,4-dihydroxyphenyl)pentanoic acid, tri-TMS	✓	✗	✓

acids coupled with terminal carbon removal (group III). A similar situation occurred at 24 h of incubation, except that the kinetics of metabolites formation decreased in *M. acridum* and none of the groups of derivatives was distinguished in this strain, which seemed to be similar to *M. globosum* (Fig. 4C). In the PCA chart (Fig. 4), *M. anisopliae* was highly differentiated due to the high activity in the formation of metabolites belonging to groups I, III, and IV.

### 3.4. Effect of 4-*n*-NP on CYP450 activity

In all the tested strains, except in *M. lepidiotae* ( $P > 0.05$ ), 4-*n*-NP caused an increase in the activity of CYP450 in comparison to the respective biotic controls (Fig. 5). It was also observed that its activity in biotic controls differed slightly among the tested strains. *M. brunneum* and *M. globosum* exhibited the highest CYP450 activity in the samples supplemented with 4-*n*-NP compared to appropriate controls without toxic substrate. The activity where 3.4 and 4.8-fold respectively for *M. brunneum* and *M. globosum* ( $P < 0.05$ ) (Fig. 5). In other strains, except of *M. lepidiotae*, the activity of the tested protein in the samples supplemented with 4-*n*-NP was at least doubled ( $P < 0.05$ ) (Fig. 5).

Numerous filamentous fungi are characterized by the presence of CYP450, which, due to its versatile catalytic properties, is involved in various metabolic processes, e.g., in the removal of xenobiotics (Durairaj et al., 2016). An example may be its participation in the elimination of dibutyltin (DBT) by the microscopic fungus *M. robertsii* (Siewiera et al., 2017), the elimination and detoxification of 2,4-dichlorophenoxyacetic acid (2,4-D) by the filamentous fungus *U. isabellina* (Nykiel-Szymańska et al., 2018b), phenanthrene degradation by *Pleurotus ostreatus* (Bezalel et al., 1996) or by *Cunninghamella elegans* (Cerniglia and Yang, 1984), and its involvement in the transformation and detoxification of alachlor by *Trichoderma koningii* (Nykiel-Szymańska et al., 2018a). CYP450 monooxygenases have also been described in fungi belonging to the genus *Metarhizium* in terms of their contribution to the

degradation and penetration of insect cuticle rich in alkanes and hydrocarbons, which is the first barrier in the protection against infection (Lin et al., 2011). Among the CYP450 genes, the family CYP52 occurs in *Metarhizium* sp. and is involved in the first hydroxylation stage in the process of assimilation of alkanes and the CYP53 clan, related to the metabolism and detoxification of aromatic hydrocarbons. According to the literature data, *M. robertsii*, as well as *M. anisopliae*, have two genes of both the CYP52 and CYP53 families (Huarte-Bonnet et al., 2017). In connection with the evidence for an increase in CYP450 activity in the cultures with the addition of 4-*n*-NP, an experiment to reveal the role of this enzyme during the process of xenobiotics degradation was carried out. The *M. robertsii* strain was selected as a research model, as a previous study has already confirmed CYP450 activity (Huarte-Bonnet et al., 2017). Moreover, our previous results (Różalska et al., 2015b) and also data obtained in this study indicate that *M. robertsii* underwent hydroxylation of 4-*n*-NP at many different positions within the first few hours of incubation, which suggests the participation of CYP450 in the degradation of 4-*n*-NP.

The results indicate that the role of the CYP is crucial for the initiation of degradation of 4-*n*-NP by *M. robertsii*. The addition of ABT at 0 h of incubation inhibited the entire pathway (Tab. 2). Among the derivatives of 4-*n*-NP degradation, we paid special attention to the metabolites from groups I, III, and IV, because these were detected in samples at 6 h of incubation (Fig. 4, Fig. 3S). In the samples with the addition of ABT at 0 h of incubation, the presence of derivatives such as Phenylacetic acid, TMS (**1**); 4-Hydroxybenzoic acid, di-TMS (**3**); and 2-(4-Hydroxyphenyl)acetic acid, di-TMS (**4**) was found, but most likely they were not products of xenobiotic decomposition, but were derived from fungal cells similar to what we had suggested in our previous work (Tab. 2) (Różalska et al., 2015b). The formation of metabolites was not inhibited when ABT was added after 6 h of incubation.

CYP450 is involved in the hydroxylation of 4-*n*-NP at the end of the aliphatic chain and also at the aromatic ring. This has been confirmed in salmon hepatic microsomes, where hydroxylation

occurred at various positions and several CYP450 enzymes were involved in this process. The 9-(4-Hydroxyphenyl)nonanoic acid, di-TMS as well as 8-(4-Hydroxyphenyl)octanoic acid, di-TMS and 7-(4-Hydroxyphenyl)heptanoic acid, di-TMS were identified as the main metabolites (Thibaut et al., 2002). Moreover, in the elimination of NP by human liver cells, CYP450 was involved in the process of hydroxylation at the 1' carbon position, identifying the metabolite 4-(1-Hydroxynonyl)phenol, di-TMS (Deng et al., 2010). In transgenic tobacco cell cultures, CYP450 enzymes were involved in hydroxylation of 4-*n*-NP at the 4', 5', 6', 7', and 8' carbon positions of the aliphatic chain as well as in the hydroxylation of the aromatic ring (Berger et al., 2005).

#### 4. Conclusion

The genus *Metarhizium* has the characteristic feature of the biodegradation of 4-*n*-NP. For the first time, a xenobiotic degradation pathway, which is distinctive for this genus has been proposed. The pathway is complex and consists of four coexisting 4-*n*-NP degradation routes with versatile hydroxylations in the alkyl chain and in the aromatic ring. Moreover, the role of CYP450 in the elimination of 4-*n*-NP by *M. robertsii* has been confirmed.

#### Acknowledgments

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#### Appendix A. Supplementary data

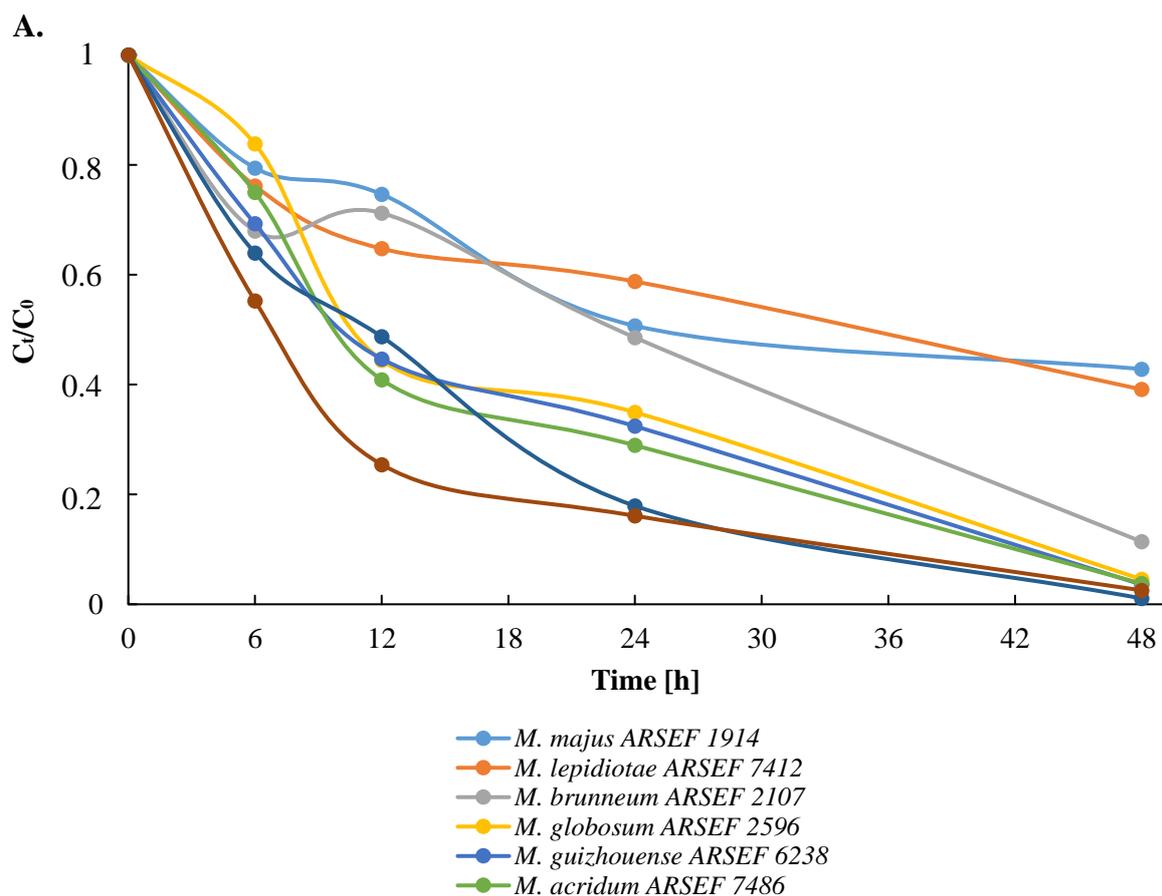
E-supplementary data of this work can be found in the online version of the paper.

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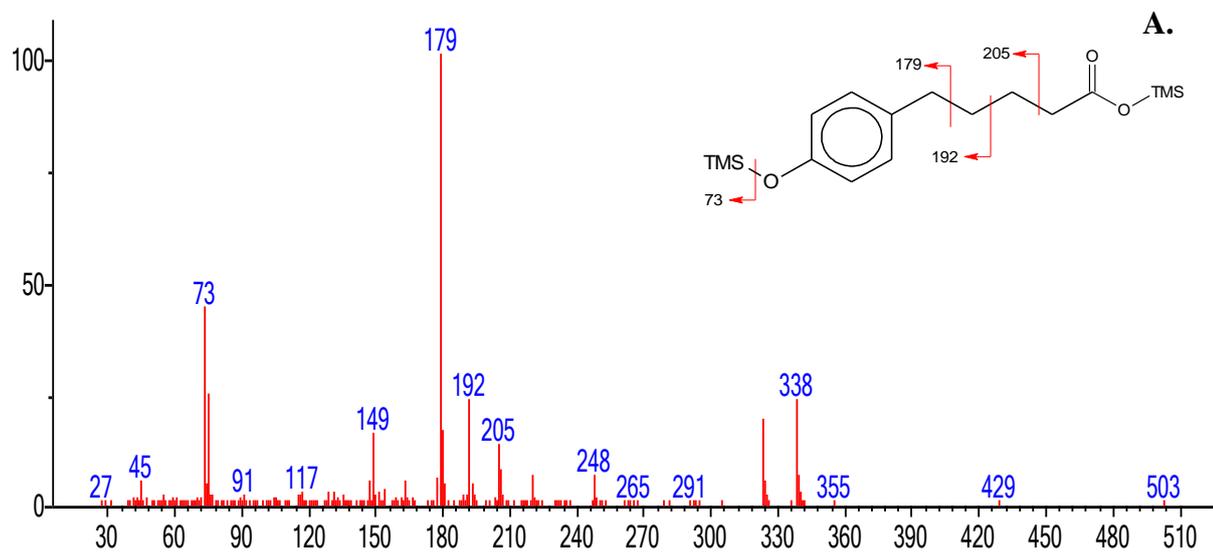
Supplementary material



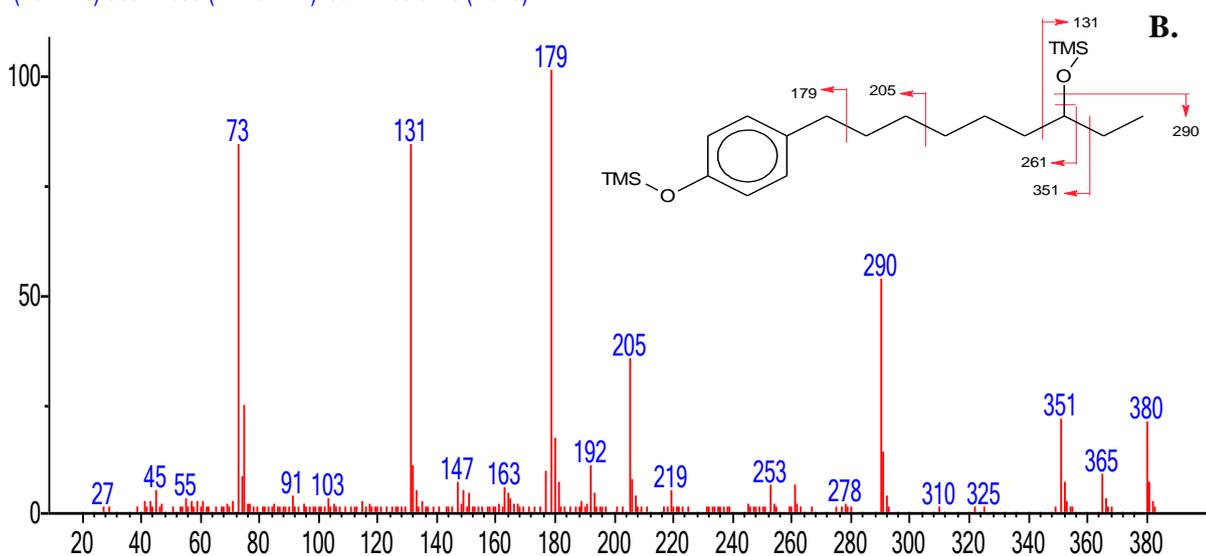
**Figure 1S.** Kinetics of degradation of 4-*n*-nonylphenol (4-*n*-NP) by *Metarhizium* sp.

**A.** Removal rate analysis  $C_t/C_0$  ( $C_t$  = concentration at time  $t$ ;  $C_0$  = initial concentration);

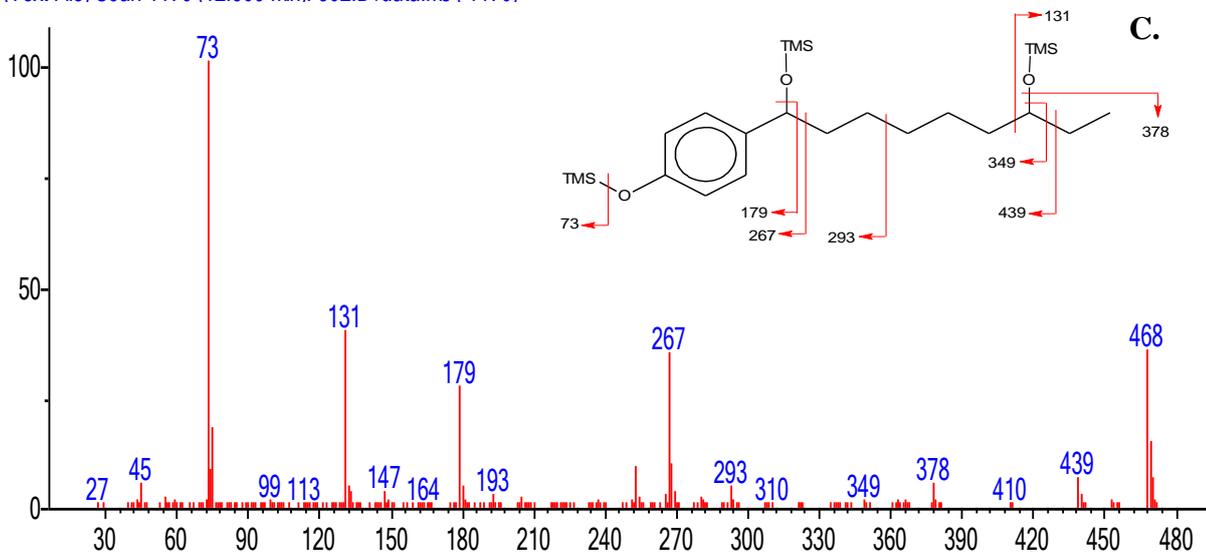
**B.** Degradation rate constant ( $k$ ).



(Text File) Scan 1030 (11.464 min): 307.D\data.ms (-1026)

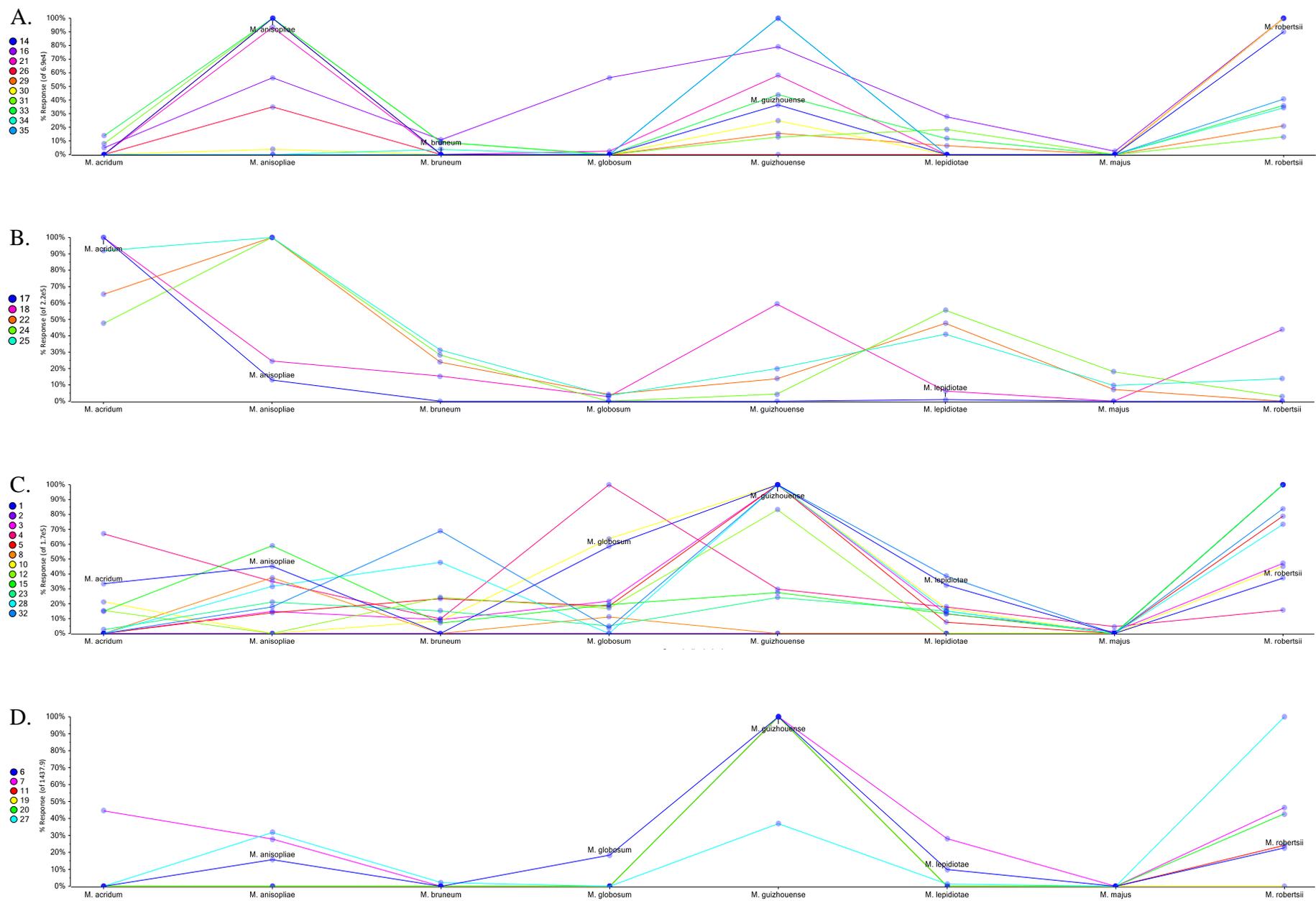


(Text File) Scan 1176 (12.509 min): 302.D\data.ms (-1170)

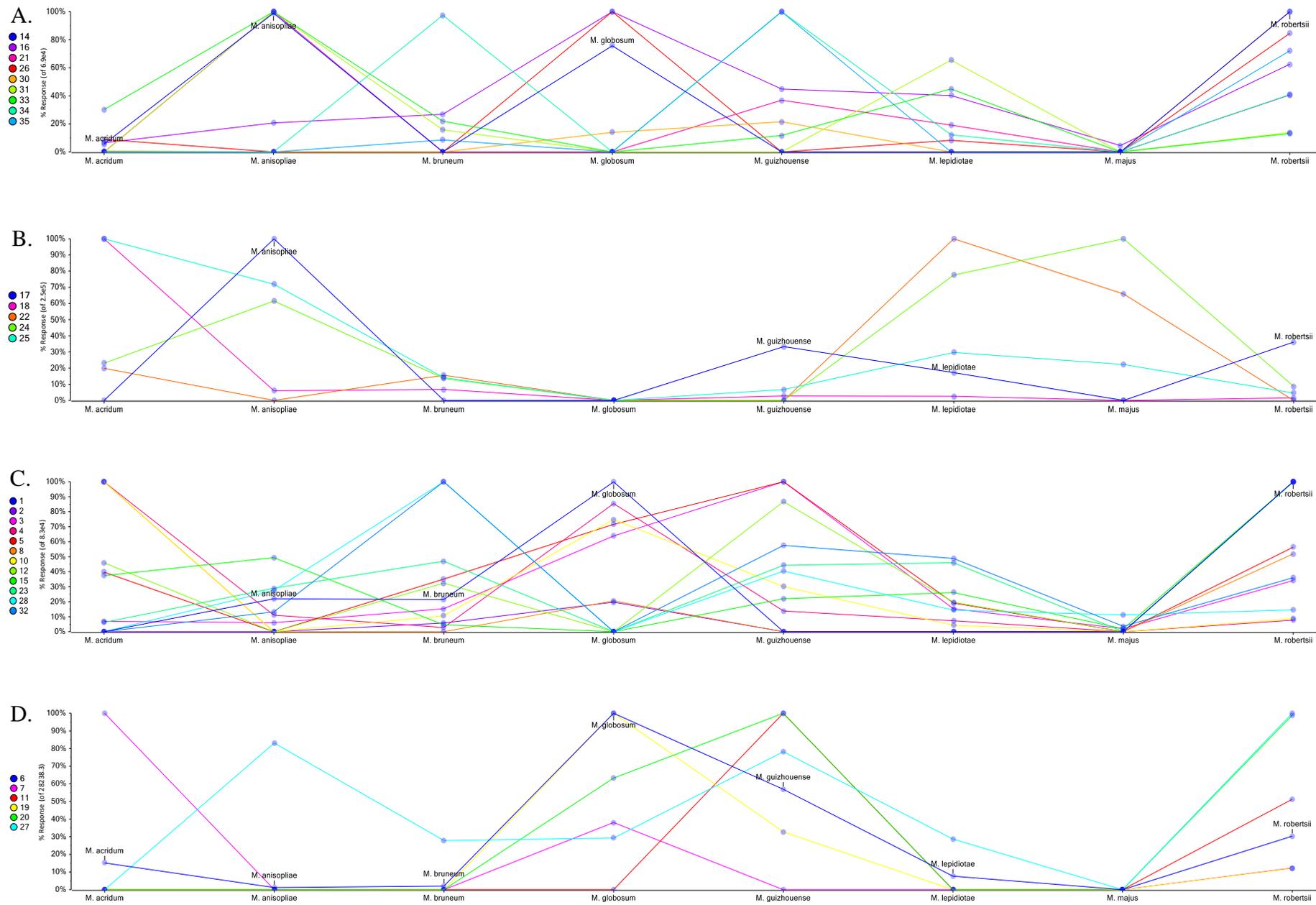


(Text File) Scan 1278 (13.240 min): 302.D\data.ms (-1275)

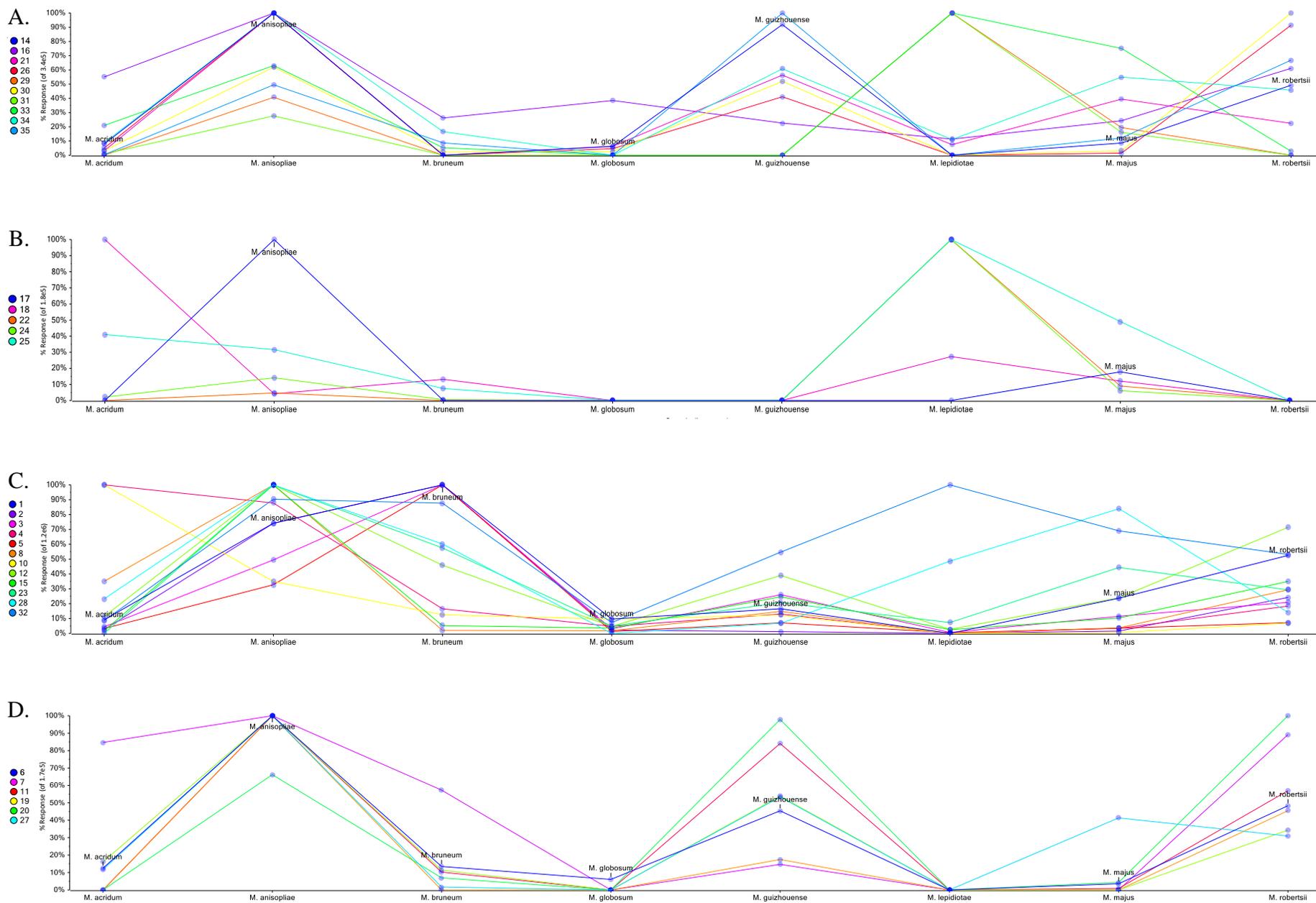
**Figure 2S.** Mass spectrum analysis of selected 4-*n*-NP intermediates. **A.** 4-(1-Hydroxynonyl)phenol, di-TMS **B.** (4-Hydroxyphenyl)octanoic acid, di-TMS **C.** 9-Hydroxy-9-(4-hydroxyphenyl)nonanoic acid, tri-TMS



**Figure 3S.** Plots of peaks responses of four metabolites groups (A – group I, B– group II, C– group III, D – group IV) obtained from *Metarhizium* sp. cultures extracts after 6 h of incubation



**Figure 4S.** Plots of peaks responses of four metabolites groups (A – group I, B– group II, C– group III, D – group IV) obtained from *Metarhizium* sp. cultures extracts after 12 h of incubation



**Figure 5S.** Plots of peaks responses of four metabolites groups (A – group I, B– group II, C– group III, D – group IV) obtained from *Metarhizium* sp. cultures extracts after 24 h of incubation